

Notes

**Base-Induced Bridge Cleavage of
1,5-Dimethyl-7-oxabicyclo[2.2.1]hept-5-ene
Derivatives: A Short Synthesis of an
Advanced Intermediate for the A and C
Ring Subunits of Taxol**

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Since the isolation of taxol, this tetracyclic diterpene has attracted tremendous attention due to its great potential in the successful treatment of several types of cancer.¹ On the other hand, the structural complexity of this molecule has stimulated general elegant synthetic approaches,² including four total syntheses.³

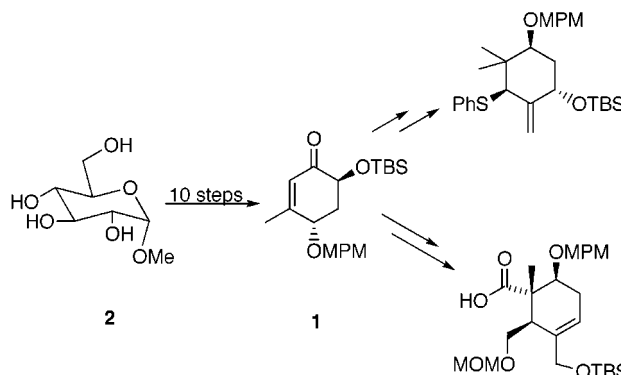
Among the disconnective approaches to the tetracyclic ring systems of taxol,⁴ we have focused our attention on two papers of Ermolenko et al.⁵ in which the ketone **1** constitutes an advanced intermediate for the synthesis of A and C ring subunits (Scheme 1).

(1) For some selected references on the biological relevance of taxol, see: (a) Georg, G. I.; Chem, T. T.; Ojima, I.; Vyas, M. D. *Taxane Anticancer Agents* American Chemical Society: Washington, 1995. (b) Kinston, D. G. L.; Molinero, A. A.; Rimoldo, J. M. *Progress in the Chemistry of Organic Natural Products*; Springer-Verlag: New York, 1993.

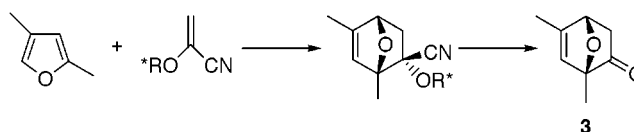
(2) (a) Swindell, C. S. *Studies in Natural Products Chemistry*; Elsevier: New York, 1993; Vol. 12. (b) Boa, A. N.; Jenkins, P. R.; Lawrence, N. J. *Recent Progress in the Synthesis of Taxanes*; Contemporary Organic Synthesis I. Royal Society of Chemistry; Vol. 1, pp 47–75. (c) Nicolaou, K. C.; Dai, W. M.; Guy, R. K. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 15.

(3) (a) Nicolaou, K. C.; Yang, Z.; Liu, J. J.; Ueno, H.; Nantermet, P. G.; Guy, R. K.; Claiborne, C. F.; Renaud, J.; Coulaudouros, E. A.; Paulvannan, K.; Sorensen, E. J. *Nature* **1994**, *367*, 630. (b) Holton, R. A.; Somoza, C.; Kim, H. B.; Liang, F.; Biediger, R. J.; Boatman, P. D.; Shindo, M.; Smith, C. C.; Kim, S.; Nadizadeh, H.; Suzuki, Y.; Tao, C.; Vu, P.; Tang, S.; Zhang, P.; Murthi, K. K.; Gentile, L. N.; Liu, J. H. *J. Am. Chem. Soc.* **1994**, *116*, 1597. (c) Holton, R. A.; Somoza, C.; Kim, H. B.; Liang, F.; Biediger, R. J.; Boatman, P. D.; Shindo, M.; Smith, C. C.; Kim, S.; Nadizadeh, H.; Suzuki, Y.; Tao, C.; Vu, P.; Tang, S.; Zhang, P.; Murthi, K. K.; Gentile, L. N.; Liu, J. H. *J. Am. Chem. Soc.* **1994**, *116*, 1599. (d) Nicolaou, K. C.; Nantermet, P. G.; Ueno, H.; Guy, R. K.; Coulaudouros, E. A.; Sorensen, E. J. *J. Am. Chem. Soc.* **1995**, *117*, 624. (e) Nicolaou, K. C.; Liu, J. J.; Yang, Z.; Ueno, H.; Guy, R. K.; Claiborne, C. F.; Sorensen, E. J.; Hwang, C. K.; Nakada, M.; Nantermet, P. G. *J. Am. Chem. Soc.* **1995**, *117*, 634. (f) Nicolaou, K. C.; Yang, Z.; Liu, J. J.; Nantermet, P. G.; Claiborne, C. F.; Renaud, J.; Guy, R. K.; Shibayama, K. *J. Am. Chem. Soc.* **1995**, *117*, 645. (g) Nicolaou, K. C.; Nantermet, P. G.; Ueno, H.; Liu, J. J.; Yang, Z.; Renaud, J.; Paulvannan, K.; Chadha, R. *J. Am. Chem. Soc.* **1995**, *117*, 653. (h) Danishefsky, S. J.; Masters, J. J.; Young, W. B.; Link, J. T.; Snyder, L. B.; Magee, T. V.; Jung, D. K.; Isaacs, R. C. A.; Bornmann, W. G.; Alaimo, C. A.; Coburn, C. A.; Di Grandi, M. J. *J. Am. Chem. Soc.* **1996**, *118*, 2843. (i) Wender, P. A.; Badham, N. F.; Conway, S. P.; Floreancig, P. E.; Glass, T. E.; Granicher, C.; Houze, J. B.; Janichen, J.; Lee, D.; Marquess, D. G.; McGrane, P. L.; Meng, W.; Mucciario, T. P.; Muhlebach, M.; Natchus, M. G.; Paulsen, H.; Rawlins, D. B.; Satkofsky, J.; Shuker, A. J.; Sutton, J. C.; Taylor, R. E.; Tomooka, K. *J. Am. Chem. Soc.* **1997**, *119*, 2755. (j) Wender, P. A.; Badham, N. F.; Conway, S. P.; Floreancig, P. E.; Glass, T. E.; Houze, J. B.; Krauss, N. E.; Lee, D.; Marquess, D. G.; McGrane, P. L.; Meng, W.; Natchus, M. G.; Shuker, A. J.; Sutton, J. C.; Taylor, R. E. *J. Am. Chem. Soc.* **1997**, *119*, 2757. (k) See also: Mukaiyama, T. *Aldrichimica Acta* **1996**, *29*, 59.

Scheme 1



Scheme 2



The ketone **1** was prepared by Ermolenko et al. in 10 steps from methyl- α -D-glucopyranoside **2**.

In this paper, we wish to account for the synthesis of an analogue of **1** using as a key step a new methodology,⁶ developed for us, for the ring opening of 1,5-dimethyl-7-oxanorbornenone **3**. Compound **3** has been previously synthesized by Vogel et al. in optically pure form⁷ in two steps, starting from 2,4-dimethylfuran (available in his turn from mesityl oxide in two steps⁸) and 2-canfanoxo-acrilonitrile (Scheme 2).

The reaction of compound **3** with NaBH₄ in MeOH afforded alcohol **4**. When **4** was treated with LDA in a

(4) See ref 2b. For some recent references concerning the synthesis of the A and C ring subunits, see: (a) Grimaud, L.; Ferezou, J. P.; Prunet, J.; Lallemand, J. Y. *Tetrahedron* **1997**, *53*, 9253. (b) Dudones, J. D.; Sampson, P. *J. Org. Chem.* **1997**, *62*, 7508. (c) Crich, D.; Natarajan, S.; Crich, J. Z. *Tetrahedron* **1997**, *53*, 7139. (d) Crich, D.; Jiao, X. Y.; Bruncko, M. *Tetrahedron* **1997**, *53*, 7127. (e) Kusama, H.; Mori, T.; Mitani, I.; Kashima, H.; Kuwajima, I. *Tetrahedron Lett.* **1997**, *38*, 4129. (f) Takahashi, T.; Iwamoto, H. *Tetrahedron Lett.* **1997**, *38*, 2483. (g) Yadav, J. S.; Sasmal, P. K. *Tetrahedron Lett.* **1997**, *38*, 8769. (h) Delalogue, F.; Prunet, J.; Pancrazi, A.; Lallemand, J. Y. *Tetrahedron Lett.* **1997**, *38*, 237. (i) Magnus, P.; Tavares, F.; Westwood, N. *Tetrahedron Lett.* **1997**, *38*, 1341. (j) Hirai, Y.; Suga, T.; Nagaoka, H. *Tetrahedron Lett.* **1997**, *38*, 4997. (k) Yadav, J. S.; Srinivas, D. *Tetrahedron Lett.* **1997**, *38*, 7789. (l) Stork, G.; Doi, T.; Liu, L. *Tetrahedron Lett.* **1997**, *38*, 7471. (m) Nakada, M.; Kojima, E.; Iwata, Y. *Tetrahedron Lett.* **1998**, *39*, 313. (n) Stork, G.; Manabe, K.; Liu, L. *J. Am. Chem. Soc.* **1998**, *120*, 1337.

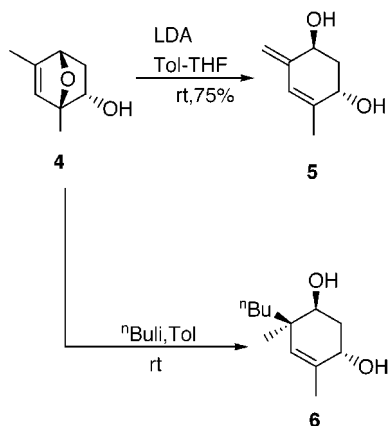
(5) (a) Ermolenko, M. S.; Shekharam, T.; Lukacs, G.; Potier, P. *Tetrahedron Lett.* **1995**, *36*, 2461. (b) Ermolenko, M. S.; Lukacs, G.; Potier, P. *Tetrahedron Lett.* **1995**, *36*, 2465.

(6) For a preliminary account of this methodology, see: Arjona, O.; Conde, S.; Plumet, J.; Viso, A. *Tetrahedron Lett.* **1995**, *36*, 6157. For a related paper, see: Lautens, M.; Ma, S. *Tetrahedron Lett.* **1996**, *37*, 1727.

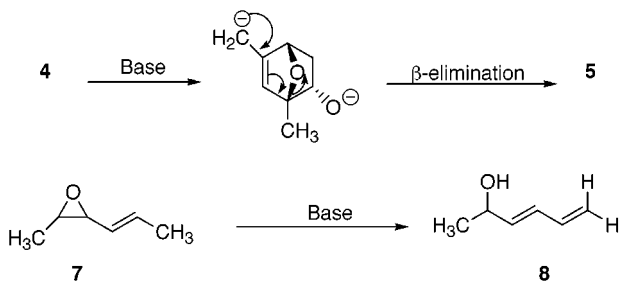
(7) (a) Kernen, P.; Vogel, P. *Helv. Chim. Acta* **1995**, *78*, 301. (b) Bialecki, M.; Vogel, P. *Helv. Chim. Acta* **1995**, *78*, 325. (c) Sevin, A. F.; Vogel, P. *J. Org. Chem.* **1994**, *59*, 5920. Although ketone **3** is available in optically pure form, all substrates used in this work are racemic.

(8) (a) Morel, T.; Verkade, P. E. *Recl. Trav. Chim. Pays-Bas* **1949**, *68*, 619. (b) Morel, T.; Verkade, P. E. *Recl. Trav. Chim. Pays-Bas* **1951**, *70*, 35.

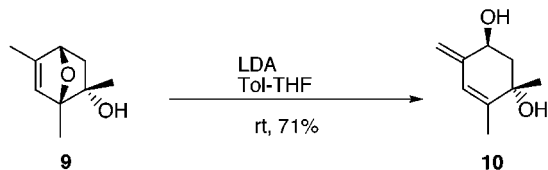
Scheme 3



Scheme 4



Scheme 5



toluene/THF mixture at room temperature, 75% yield of compound **5** was obtained (Scheme 3). The use of these reaction conditions appears to be critical for the success of the process. For instance, when THF alone is used as a solvent, the isolated yield falls to 53%. With $n\text{BuLi}$ as a basic reagent, a mixture of **5** (25%), compound **6**⁹ (25%), and starting material (50%) was observed as reaction crude (Scheme 3).

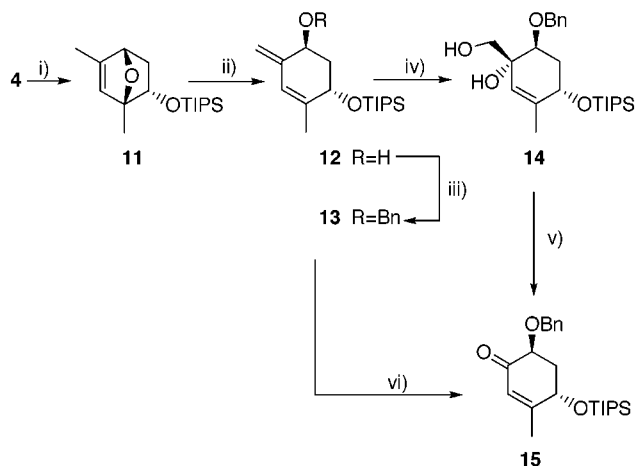
The structure of **5** was confirmed from its spectroscopic data. Thus, in the ^1H NMR spectrum (300 MHz), three singlets were observed corresponding to the three vinylic protons at 5.89, 5.24, and 5.05 ppm. The protons attached to the two hydroxy centers appear as two multiplets at 4.50–4.36 and 4.30–4.17 ppm. Finally, the methyl group gives an apparent singlet at 1.82 ppm.

From the mechanistic point of view, this process should occur by deprotonation of the allylic methyl group followed by β -elimination of the oxygen bridge in a reaction, reminiscent of the well-known transformation of acyclic vinyloxiranes in the related dienols,¹⁰ i.e., transformation from **7** to **8** (Scheme 4).

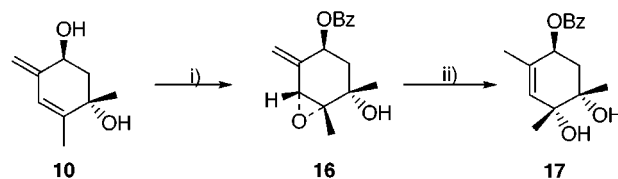
Some derivatives of **4** show the same behavior. For instance, let us take compound **9**, which was obtained

(9) Compound **6** arises from the alkylative ring opening of the starting bicyclic by attacking with organolithium reagent position 5 of compound **4**. For leading references, see: Arjona, O.; Fernandez de la Pradilla, R.; Garcia, E.; Martin, A.; Plumet, J. *Tetrahedron Lett.* **1989**, *30*, 6437. (b) Arjona, O.; Fernandez de la Pradilla, R.; Martin, A.; Plumet, J. *Tetrahedron* **1990**, *46*, 8199.

(10) Thummel, R. P.; Rickborn, B. *J. Org. Chem.* **1972**, *37*, 4250.

Scheme 6^a

^a (i) TIPSOTf, Et_3N , 4-DMPA, CH_2Cl_2 , 0°C , 89%; (ii) LDA, Tol/THF, rt, 87%; (iii) BnBr, NaH, THF, rt, 83%; (iv) OsO_4 , acetone– H_2O 8:1, rt, 60%; (v) NaIO_4 , THF/ H_2O 1:1, rt, 100%; (vi) NaIO_4 , $\text{RuCl}_3\cdot\text{H}_2\text{O}$, $\text{CH}_3\text{CN}/\text{CCl}_4/\text{H}_2\text{O}$ 1:1:1.4, rt, 56%.

Scheme 7^a

^a (i) a) BzCl , Et_3N , CH_2Cl_2 , 0°C , 70%. b) $^t\text{BuOOH}$, $\text{VO}(\text{acac})_2$, C_6H_6 , rt, 95%. (ii) H_2 , PtO_2 , NaNO_2 , MeOH, rt, 60%.

from **3** by reaction with MeMgBr . This compound, under treatment with LDA (Tol/THF, rt, 71%) gives the expected product **10** (Scheme 5).

Application of this methodology to the synthesis of compounds related to **1** has been achieved in the following way: reaction of **4** with TIPSOTf affords **11**, which in turn gives **12** by the previously described ring-opening methodology. Benzoylation of **12** followed by osmium tetroxide oxidation of the resulting protected diol **13** gives compound **14**. Finally, periodate cleavage of **14** affords **15**, an analogue of **1**. It should be pointed out that the transformation **13** to **15** can be achieved in a single step by reaction of **13** with $\text{NaIO}_4/\text{RuCl}_3$ ¹¹ (Scheme 6).

Another useful transformation based on the above-described ring-opening procedure afforded the synthesis of cyclohexene derivatives with three chiral centers (two quaternary carbon atoms) possessing a well-defined stereochemistry. Thus, benzoylation of **10** followed by selective epoxidation of the endocyclic double bond ($^t\text{BuOOH}/\text{C}_6\text{H}_6$, $\text{VO}(\text{acac})_2$, 95%)¹² regioselectively afforded epoxide **16**. This was transformed into the monoprotected triol **17** by hydrogenation, using $\text{PtO}_2/\text{NaNO}_2$ in MeOH anhyd.,¹³ with concomitant double bond migration. (Scheme 7).

In summary, a new efficient protocol for the ring opening of 1,5-dimethyl-7-oxanorbornene derivatives has been described. This methodology has been applied to the

(11) Carlsen, H. J.; Katsuki, T.; Martin, V. S.; Sharpless, K. B. *J. Org. Chem.* **1981**, *46*, 3936.

(12) (a) Rao, A. S. In *Comprehensive Organic Synthesis*; Trost, B. M., Ed.; Pergamon Press: Oxford, 1984; Vol. 7, Chapter 3, p 357. (b) Itoh, T.; Jitsukawa, K.; Kaneda, K.; Teranishi, S. *J. Am. Chem. Soc.* **1979**, *101*, 159.

(13) Dart, M. C.; Henbest, H. B. *J. Chem. Soc.* **1960**, *82*, 3563.

preparation of an advanced synthetic intermediate for the synthesis of the A and C ring subunits of taxol starting from the ketone **3** (six steps, 36% overall yield, or five steps, 34% overall yield). Moreover, the stereocontrolled synthesis of methyl-substituted cyclohexene derivatives has been also reported.

Experimental Section

General Methods. Reagents and solvents were handled by using standard syringe techniques. Tetrahydrofuran and diethyl ether were distilled over sodium and benzophenone; benzene over sodium; toluene, dichloromethane, and triethylamine over calcium hydride; pyridine over KOH, and acetone over KMnO_4 . The remaining solvents and chemicals were commercial and used as received. ^1H NMR and ^{13}C NMR were recorded at 250 or 300 MHz. Flash chromatography was performed using 230–400 mesh silica gel. Analytical TLC was carried out on silica gel plates. Melting points are uncorrected. Elemental analyses were performed at the Universidad Complutense de Madrid. Compound **3** was prepared according with the method of Vogel et al.⁷

(2S*)-1,5-Dimethyl-7-oxabicyclo[2.2.1]hept-5-en-2-ol, 4. To a solution of **3** (348 mg, 2.52 mmol) in MeOH (4 mL), NaBH_4 (143 mg, 3.78 mmol) was added at -20°C . The mixture was allowed to reach room temperature while stirring for 2 h, and then water was added. The aqueous layer was extracted with AcOEt, dried over MgSO_4 , and concentrated in vacuo. **4** was obtained after purification by column chromatography (hexane/AcOEt 2:1) as a colorless oil (332 mg, 94%). Data for **4**: ^1H NMR (CDCl_3 , 250 MHz) δ 5.75 (d, 1H, $J = 1.6$ Hz), 4.54 (d, 1H, $J = 4.9$ Hz), 3.94–4.09 (m, 1H), 2.41 (ddd, 1H, $J = 12.1, 8.0, 4.9$ Hz), 1.90 (d, 3H, $J = 1.6$ Hz), 1.56 (s, 3H), 1.04 (dd, 1H, $J = 12.1, 2.6$ Hz); ^{13}C NMR (62.5 MHz) δ 150.6, 127.8, 87.3, 82.1, 76.3, 38.6, 16.8, 13.1; IR (CHCl_3) ν 3420, 2980, 2940, 1450, 1390 cm^{-1} . Anal. Calcd for $\text{C}_8\text{H}_{12}\text{O}_2$: C, 68.54; H, 8.63. Found: C, 68.37; H, 8.56.

(1S*,3S*)-6-Methyl-4-methylcyclohex-5-en-1,3-diol, 5. To a solution of **4** (70 mg, 0.50 mmol) in THF (1.3 mL) at room temperature, 6 equiv of a 1 M solution of LDA (previously formed from 1 equiv of $^i\text{Pr}_2\text{NH}$ in 0.8 mL/mmol of toluene at 0°C and 1 equiv of a 1.6 M solution of $^t\text{BuLi}$ in hexanes) were added and stirred for 2 h. The reaction was quenched with 5% HCl and extracted with AcOEt. The organic layer was dried over MgSO_4 and evaporated under reduced pressure. The resulting crude was purified by column chromatography on silica gel (hexane/AcOEt 2:1) to afford 53 mg of **5** (75%) as a white solid. Data of **5**: mp, $105\text{--}106^\circ\text{C}$; ^1H NMR (CDCl_3 , 250 MHz) δ 5.89 (s, 1H), 5.24 (s, 1H), 5.05 (s, 1H), 4.36–4.50 (m, 1H), 4.17–4.30 (m, 1H), 1.82 (s, 3H), 1.58–1.74 (m, 2H); ^{13}C NMR (CDCl_3 , 62.5 MHz) δ 145.6, 138.6, 125.6, 110.8, 68.4, 67.2, 40.3, 20.2; IR (CHCl_3) ν 3300, 3020, 2920, 1420 cm^{-1} . Anal. Calcd for $\text{C}_8\text{H}_{12}\text{O}_2$: C, 68.54; H, 8.63. Found: C, 68.47; H, 8.39.

(2S*)-1,5-Dimethyl-2-methyl-7-oxabicyclo[2.2.1]hept-5-en-2-ol, 9. MeMgBr 3 M in Et₂O (1.33 mL) was added dropwise to a solution of **3** (276 mg, 2 mmol) in THF (5 mL) at 0°C . The mixture was stirred for 2 h, quenched with NH_4Cl and H_2O , and extracted with AcOEt. The organic layer was dried over MgSO_4 and evaporated under reduced pressure. **9** (234 mg, 76%) was obtained as a colorless oil after purification by column chromatography (hexane/AcOEt 4:1). Data of **9**: ^1H NMR (CDCl_3 , 250 MHz) δ 5.79 (d, 1H, $J = 1.7$ Hz), 4.52 (d, 1H, $J = 4.9$ Hz), 2.04 (dd, 1H, $J = 4.9, 12.5$ Hz), 1.88 (d, 3H, $J = 1.7$ Hz), 1.62 (s, 1H), 1.43 (s, 3H), 1.40 (s, 3H), 1.41 (d, 1H, $J = 12.5$ Hz); ^{13}C NMR (CDCl_3 , 62.5 MHz) δ 149.4, 124.3, 88.9, 81.5, 78.4, 44.9, 24.6, 13.1, 12.8; IR (CH_2Cl_2) ν 3440, 2980, 1450, 1380 cm^{-1} . Anal. Calcd for $\text{C}_9\text{H}_{14}\text{O}_2$: C, 70.10; H, 9.15. Found: C, 70.02; H, 8.93.

(1S*,3S*)-1,6-Dimethyl-4-methylcyclohex-5-en-1,3-diol, 10. To a solution of **9** (64 mg, 0.42 mmol) in THF (1.1 mL) at room temperature, 6 equiv of a 1 M solution of LDA (previously formed from 1 equiv of $^i\text{Pr}_2\text{NH}$ in 0.8 mL/mmol of toluene at 0°C and 1 equiv of a 1.6 M solution of $^t\text{BuLi}$ in hexanes) were added and stirred for 2 h. The reaction was quenched with 5% HCl and extracted with AcOEt. The organic layer was dried over MgSO_4 and evaporated under reduced pressure. The resulting crude was purified by column chromatography on silica gel

(hexane/AcOEt 2:1) to afford 50 mg of **10** (78%) as a white solid. Data of **10**: mp, $105\text{--}107^\circ\text{C}$; ^1H NMR (CDCl_3 , 300 MHz) δ 5.89 (s, 1H), 5.13 (s, 1H), 4.90 (s, 1H), 4.44–4.54 (m, 1H), 2.70–2.80 (m, 1H), 2.45–2.53 (m, 1H), 2.09 (dd, 1H, $J = 4.4, 13.0$ Hz), 1.80 (s, 3H), 1.72 (dd, 1H, $J = 11.7, 13.0$ Hz), 1.33 (s, 3H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 146.1, 139.8, 126.1, 109.0, 71.5, 66.6, 47.1, 27.7, 17.4; IR (CHCl_3) ν 3550, 2990, 2960, 2930, 1450, 1380 cm^{-1} . Anal. Calcd for $\text{C}_9\text{H}_{14}\text{O}_2$: C, 70.08; H, 9.16. Found: C, 69.97; H, 8.92.

(2S*)-1,5-Dimethyl-2-triisopropylsilyloxy-7-oxabicyclo[2.2.1]hept-5-ene, 11. To a solution of **4** (340 mg, 2.43 mmol) in CH_2Cl_2 (8 mL) at 0°C , Et_3N (0.37 mL), DMAP (0.01 equiv) and TIPSOTf (1.1 equiv) were added. The mixture was stirred for 1 h. The reaction was diluted with CH_2Cl_2 , and the organic layer was washed with 5% NaHCO_3 and NaCl, dried over MgSO_4 , and concentrated under reduced pressure. The crude product was purified by column chromatography on silica gel (hexane/AcOEt 20:1) to afford 640 mg (89%) of **11** as a colorless oil. Data of **11**: ^1H NMR (CDCl_3 , 300 MHz) δ 5.67 (d, 1H, $J = 1.2$ Hz), 4.52 (d, 1H, $J = 4.8$ Hz), 4.16 (dd, 1H, $J = 7.5, 2.4$ Hz), 2.27 (ddd, 3H, $J = 12.0, 7.5, 4.8$ Hz), 1.85 (d, 3H, $J = 1.2$ Hz), 1.54 (s, 3H), 0.91–1.17 (m, 22H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 147.0, 129.0, 87.5, 82.1, 76.7, 38.7, 17.8, 17.8, 17.8, 17.2, 12.6, 12.1; IR (CH_2Cl_2): ν 3160, 2950, 2880, 1470, 1450, 1400, 1370 cm^{-1} . Anal. Calcd for $\text{C}_{17}\text{H}_{32}\text{O}_2\text{Si}$: C, 68.86; H, 10.88. Found: C, 68.72; H, 10.78.

(1S*,5S*)-5-Triisopropylsilyloxy-4-methyl-2-methylcyclohex-3-en-1-ol, 12. To a solution of **11** (450 mg, 1.52 mmol) in THF (4 mL) at room temperature, 6 equiv of a 1 M solution of LDA (previously formed from 1 equiv of $^i\text{Pr}_2\text{NH}$ in 0.8 mL/mmol of toluene at 0°C and 1 equiv of a 1.6 M solution of $^t\text{BuLi}$ in hexanes) were added and stirred for 2 h. The reaction was quenched with 5% HCl and extracted with AcOEt. The organic layer was dried over MgSO_4 and evaporated under reduced pressure. The resulting crude was purified by column chromatography on silica gel (hexane/AcOEt 10:1) to afford 337 mg of **12** (87%) as a colorless oil. Data of **12**: ^1H NMR (CDCl_3 , 300 MHz) δ 5.82 (s, 1H), 4.97 (s, 1H), 4.82 (s, 1H), 4.38–4.50 (m, 2H), 1.86–1.95 (m, 2H), 1.81 (s, 3H), 1.50–1.55 (m, 1H), 0.93–1.15 (m, 21H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 145.9, 140.2, 124.4, 110.0, 68.8, 40.8, 20.8, 18.2, 18.2, 18.2, 12.7, 12.7, 12.7, 12.7, 12.7; IR (CHCl_3) ν 3600, 3450, 2960, 2880, 1470 cm^{-1} . Anal. Calcd for $\text{C}_{17}\text{H}_{32}\text{O}_2\text{Si}$: C, 68.86; H, 10.88. Found: C, 68.76; H, 10.80.

(1S*,5S*)-5-Benzoyloxy-1-triisopropylsilyloxy-2-methyl-4-methylcyclohex-2-ene, 13. To a NaH suspension (162 mg, 4.05 mmol) in THF (10 mL) at room temperature, a solution of **12** (300 mg, 1.01 mmol) in THF (5 mL) was added. After 10 min, BnBr (0.5 mL, 4.05 mmol) was added dropwise, followed by 0.1 equiv of Bu_4NI . The mixture was stirred for 5 h, quenched with NaCl, and extracted with Et₂O. The organic layer was dried over MgSO_4 and evaporated under reduced pressure. The crude was purified by column chromatography (hexane/AcOEt, 10:1) to afford **13** (332 mg, 83%) as a colorless oil. Data of **13**: ^1H NMR (CDCl_3 , 300 MHz) δ 7.36–7.26 (m, 5H), 5.89 (s, 1H), 4.97 (d, 2H, $J = 2.6$ Hz), 4.65 (m, 1H), 4.57 (AB system, 2H, $J_{AB} = 12$ Hz), 4.18 (dd, 1H, $J = 6.6, 2.9$ Hz), 2.23 (ddd, 1H, $J = 12.8, 6.5, 1.6$ Hz), 1.97 (ddd, 1H, $J = 12.8, 7.3, 2.9$ Hz), 1.89 (s, 3H), 1.10 (br s, 21H); ^{13}C NMR (CDCl_3 , 75 MHz) δ 144.2, 142.1, 136.9, 129.0, 128.8, 128.2, 127.7, 124.3, 111.6, 75.2, 68.7, 65.1, 38.8, 20.5, 18.2, 12.8; IR (CCl_4) ν 3080, 2980, 2880, 1650, 1470 cm^{-1} . Anal. Calcd for $\text{C}_{24}\text{H}_{38}\text{O}_2\text{Si}$: C, 74.55; H, 9.91. Found: C, 74.42; H, 9.76.

(1R*,4S*,6S*)-6-Benzoyloxy-1-hydroxymethyl-4-triisopropylsilyloxy-3-methylcyclohex-2-en-1-ol, 14. A solution of 2.5% OsO_4 in $^t\text{BuOH}$ (0.18 mL, 0.015 mmol) was added to a mixture of **13** (280 mg, 0.72 mmol) and Me_3NO (161 mg, 1.45 mmol) in acetone/ H_2O (8/1) (5 mL). The reaction was stirred for 6 h. The crude mixture was quenched with NaHSO_3 and evaporated under reduced pressure. The crude product was purified by column chromatography (hexane/AcOEt 10:1) to afford a colorless oil characterized as **14** (183 mg, 60%). Data of **14**: ^1H NMR (CDCl_3 , 300 MHz) δ 7.36–7.28 (m, 5H), 5.15 (s, 1H), 4.58 (AB system, 2H, $J_{AB} = 10.8$ Hz), 4.32 (dd, 1H, $J = 4.5, 3.6$ Hz), 4.02 (dd, 1H, $J = 4.7, 3.6$ Hz), 3.84 (dd, 1H, $J = 11.7, 3.0$ Hz), 3.30 (dd, 1H, $J = 11.7, 10.2$ Hz), 3.17 (s, 1H), 2.86 (dd, 1H, $J = 10.1, 3.1$ Hz), 2.14 (dt, 1H, $J = 12.4, 3.6$ Hz), 1.94 (dt, 1H, $J = 12.4, 4.5$ Hz), 1.80 (s, 3H), 1.09 (s, 21H); ^{13}C NMR

(CDCl₃, 75 MHz) δ 141.1, 139.0, 128.7, 128.1, 127.9, 125.5, 81.1, 72.5, 72.4, 69.1, 65.7, 34.6, 20.9, 18.3, 12.9; IR (CCl₄) ν 3500, 3020, 2980, 1650 cm⁻¹. Anal. Calcd for C₂₄H₄₀O₄Si: C, 68.53; H, 9.58. Found: C, 68.41; H, 9.42.

(4S*,6S*)-6-Benzoyloxy-4-triisopropylsilyloxy-3-methylcyclohex-2-en-1-one, 15. *Method A.* To a solution of **14** (67 mg, 0.16 mmol) in THF/H₂O 1:1 (1 mL), NaIO₄ (51 mg, 0.24 mmol) was added. The mixture was stirred for 5 h, quenched with water, and extracted with AcOEt. The organic layer was dried over MgSO₄ and evaporated under reduced pressure to afford **15** as a colorless oil (62 mg, quantitative yield).

Method B. To a well-stirred solution of **13** (49 mg, 0.13 mmol) in a mixture of CH₃CN (0.73 mL), CCl₄ (0.73 mL), and H₂O (1.04 mL), NaIO₄ (111 mg, 0.52 mmol) was added. The resulting mixture was allowed to stir for 10 min, and then RuCl₃·H₂O (1.1 mg, 0.005 mmol) was added. The mixture turned black and was stirred for 2 h. The reaction was quenched with sat. NaHCO₃ and extracted with CH₂Cl₂. The organic layer was dried over MgSO₄ and evaporated in vacuo. After purification by column chromatography on silica gel (hexane/AcOEt 6:1), **15** was obtained as a colorless oil (27 mg, 56%). Data of **15**: ¹H NMR (CDCl₃, 300 MHz) δ 7.40–7.28 (m, 5H), 5.81 (s, 1H), 4.73 (AB system, 2H, J = 10.8 Hz), 4.71 (dd, J = 5.4, 4.5 Hz, 1H), 4.14 (dd, J = 8.1, 3.9 Hz, 1H), 2.39 (m, 1H), 2.24 (m, 1H), 2.07 (s, 3H), 1.18 (s, 21H); ¹³C NMR (CDCl₃, 75 MHz) δ 198.0, 162.8, 138.1, 128.4, 127.9, 127.8, 124.9, 75.7, 72.3, 68.6, 39.4, 21.7, 18.2, 12.7; IR (CCl₄) ν 2925, 2868, 1718, 1678 cm⁻¹. Anal. Calcd for C₂₃H₃₆O₃Si: C, 71.09; H, 9.34. Found: C, 70.98; H, 9.18.

(1S*,2R*,3R*,5S*)-5-Benzoyloxy-2,3-epoxy-1,2-dimethyl-4-methylcyclohexan-1-ol, 16. To a solution of a derivative of **10**, in which the secondary alcohol is protected as benzoyloxy group (30 mg, 0.12 mmol) in C₆H₆ (1.2 mL), *t*-BuOOH was added (0.04 mL, 0.12 mmol), followed by VO(acac)₂ (3 mg, 0.01 mmol) at room temperature. The mixture was stirred for 6 h and filtered through florisil. This crude was washed with 10% Na₂S₂O₅ and sat. NaHCO₃. The aqueous layer was extracted with AcOEt. The organic layer was dried over MgSO₄, evaporated under reduced pressure, and purified by column chromatography (hexane/AcOEt 10:1) to afford **16** (31 mg, 95%) as a colorless oil. Data of **16**: ¹H NMR (CDCl₃, 300 MHz) δ 7.96 (d,

2H, J = 8.1 Hz), 7.51 (t, 1H, J = 7.2 Hz), 7.38 (t, 2H, J = 7.5 Hz), 5.73 (t, 1H, J = 3.0 Hz), 5.47 (s, 1H), 5.42 (s, 1H), 3.51 (s, 1H), 2.13 (dd, 1H, J = 4.8, 13.9), 1.77 (dd, 1H, J = 6.3, 13.9), 1.42 (s, 3H), 1.38 (s, 3H); ¹³C NMR (CDCl₃, 62.5 MHz) δ 165.6, 140.4, 133.3, 130.2, 129.7, 129.7, 128.6, 128.6, 119.7, 71.4, 68.7, 65.7, 65.2, 42.3, 26.1, 17.2; IR (CDCl₃) ν 3560, 3050, 2980, 1610, 1460 cm⁻¹. Anal. Calcd for C₁₆H₁₈O₄: C, 70.06; H, 6.61. Found: C, 69.91; H, 6.47.

(1R*,2S*,4S*)-4-Benzoyloxy-1,2,5-trimethylcyclohex-5-en-1,2-diol, 17. A mixture of **16** (52 mg, 0.19 mmol) and PtO₂/NO₂Na catalyst (1:1, 4:1 mg, 0.1 mmol) in dry MeOH (10 mL) was hydrogenated (5 psi) for 2 h. The reaction mixture was filtered through a short path of Celite with CH₂Cl₂. The solvent was evaporated under reduced pressure and the product was purified by column chromatography (CH₂Cl₂/AcOEt, 1:1) to afford **17** (32 mg, 60%) as a colorless oil. Data of **17**: ¹H NMR (CDCl₃, 300 MHz) δ 8.02 (d, 2H, J = 7.8 Hz), 7.56 (t, 1H, J = 7.2 Hz), 7.43 (t, 2H, J = 7.8 Hz), 5.73 (t, 1H, J = 6.6 Hz), 5.50 (s, 1H), 2.42 (dd, 1H, J = 6.0, 14.3 Hz), 2.41 (br s, 1H), 2.20 (br s, 1H), 1.87 (dd, 1H, J = 6.6 Hz, 14.3 Hz), 1.73 (s, 3H), 1.31 (s, 6H); ¹³C NMR (CDCl₃, 75 MHz) δ 166.1, 133.3, 132.7, 131.3, 130.4, 129.6, 129.6, 128.4, 128.4, 73.8, 72.4, 71.5, 39.5, 24.5, 24.0, 19.2; IR (CH₂Cl₂): ν 3440, 3060, 2980, 2940, 1720, 1600, 1460 cm⁻¹. Anal. Calcd for C₁₆H₂₀O₄: C, 69.54; H, 7.30. Found: C, 69.49; H, 7.19.

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Supporting Information Available: ¹H NMR spectra copies for compounds **4**, **5**, **9**, **10**, **12**–**17** (10 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.

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